

STAVUDINE GASTRO-RETENTIVE MICROSPHERES OF D-GALACTURONIC ACID SODIUM: FORMULATION, CHARACTERIZATION, *INVITRO* AND *INVIVO* EVALUATION

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ABSTRACT

The prime objective of the present study was to formulate sustained release microspheres of stavudine which will retard drug release from dosage form, reduce the frequency of drug administration, minimizing the adverse effects thereby increasing the patient compliance. Sustained release microspheres of stavudine (STV), the FDA-approved drug for the treatment of HIV infections, are administered either alone or in combination with other antiretroviral agents. STV is typically administered orally as a capsule and an oral solution. The side effects of STV are dose dependent and a reduction of the total administered dose reduces the severity of the toxicity. Microencapsulation improves the drug absorption and minimizes side effects due to the localized build up of drugs against the gastrointestinal mucosa. Sustained release microspheres were prepared by ionotropic gelation method using D-Galacturonic acid sodium (sodium alginate) and natural polymers such as guar gum, locust gum and xanthan gum. These microspheres were evaluated for yield, entrapment efficiency, and *invitro* release kinetics and *invivo* pharmacokinetic studies. FT-IR and DSC results proved that there was no interaction found between drug and polymer. Scanning electron microscopy results of optimized microspheres showed discrete, spherical microspheres. The results showed that yield, entrapment efficiency was influenced by polymer concentration and stirring speed. Results of the *invitro* study showed that the desired release rate was achieved by MF4 formulation releasing drug upto 12h.

Keywords: Stavudine, Gastro-retentive microspheres, Ionotropic gelation method, D-galacturonic acid sodium, Sodium alginate, Guar gum, Locust bean gum, Xanthan gum.

INTRODUCTION

Gastro-retentive floating microspheres are low-density systems that have sufficient buoyancy to float over gastric contents and remain in stomach for prolonged period. As the system floats over gastric contents, the drug is released slowly at desired rate resulting in increased gastric retention with reduced fluctuations in plasma drug concentration.¹ A wide range of microencapsulation techniques has been developed to date. The selection of the technique depends upon the nature of the polymer, drug, and the intended use. The physicochemical parameters to be considered during the preparation of microspheres include particle size distribution, ratio of drug to polymer, entrapment/encapsulation efficiency.²

Floating drug delivery systems (FDDS) or hydrodynamically balanced systems have a bulk density lower than gastric fluids and therefore remain floating in the stomach without affecting the gastric emptying rate for a prolonged period. The drug is slowly released at a desired rate from the floating system and after the comp

leterelease;theresidualsystemisexpelledfromthestomach.ThisleadstoanincreaseintheGRTandbettercontroloverflu ctuationsinplasmadrugconcentration.

Microencapsulationistheprotectivetechnologyofencapsulating solid, liquidorgasmaterialsintomicroparticleswithad iameterof1–

1000µm,andhasbeenwidelyusedinfieldsofmedicine,cosmetics,food,textileandadvancedmaterials.³Theuniqueadvan tageofmicroencapsulationliesinthatthecorematerialiscompletelycoatedandisolatedfromexternalenvironment.Mo reimportantly,microencapsulationwouldnotaffectthepropertiesofcorematerials,providedthatpropershellmaterialan dpreparingmethodarechosen.⁴Microspheresofferanumberofadvantagesintherapeutics.Microspheresareofthedrugd eliverysystemwhichprovideprogrammedandcontrolledreleasedrugafterproperdurationofactionatparticularsite.⁵D- Galacturonicacidsodium(sodiumalginate,SA)isnaturalpolymersarethattheyarebiocompatible,biodegradableandpr oducenosystemictoxicityonadministration.Numeroushydrophilicpolymers,andinparticularpolysaccharides,aswell astheirderivatives,havebeenproposedfortheformulationofmodified- releasedosageforms.Alginates,naturalhydrophilicpolysaccharidederivedfromseaweed,consistof1→4,linkedD- mannuronicacidandL- glucuronicacidresiduesarrangedasblocksofeithertypeofunitorasarandomdistributionofeachtype.Alginatesdonotgel sincetheyhavepoly(L- glucuronicacids)whicharerigid.Alginatesareeasilygelledinpresenceofadivalentcationascalciumion.Thegelationorcro sslinkingisduetothe stackingoftheglucuronicacidblocksofalginatechains.

Stavudine(STV)isaFDA-approveddrugusedforthetreatmentofHIVinfections,AIDSandAIDS- relatedconditionsadministeredeitheraloneorincombinationwithotherantiretro- viralagent.STVisanucleosidereversetranscriptaseinhibitoranalogofthymidine.STVhasbeenrecommendedtounderg ophase-outmanagementduetoitslong-term,irreversible side-effects.STVtriphosphateinhibitstheactivityofHIV- 1reversetranscriptase(RT)bycompetingwiththenaturalsubstratethymidinetriphosphateandbycausingDNAchainter minationfollowingitsincorporationintoviralDNA.STVtriphosphateinhibitscellularDNA polymerasesβandγmarked lyreducesthesynthesisofmitochondrialDNA.STVmicrosphereshavebeenformulatedearlier.⁶ ⁸Srikrishna^{et al.},employedemulsiongelationtechnique topreparemultiparticulatefloatingformulationsofSTV.⁹STV beadswereformulatedusingcombinationofD- Galacturonicacidsodium,HPMCandethylcellulose.¹⁰HoweverpastresearchshowsnoworkonSTVmicrospheresmad eusing sodiumalginate(SA)incombinationwithguargum(GG),locustbeangum(LG)andxanthangum(XG).

The main objective of the floating microspheres of STV were prepared by using ionic gelation method with an aim of increa sing the gastric residence time and for controlled release and increase the bioavailability of the formulation. Sodium alginat e, polymeric mixture of sodium alginate in combination with guargum, locust bean and xanthangum were used as poly me ric blend to prepare the microspheres. Sodium bicarbonate was used as the gas-forming agent.

MATERIALS AND METHODS

Stavudine purchased from Suryanarayana Pharmaceuticals suppliers, Hyderabad, India. Guargum (GG), locust bean gum (LG) and xanthan gum (XG) were obtained from Signet Chemical Corporation, Mumbai, India. D- Galacturonic acid sodium (sodium alginate) from Merck Specialities Pvt Ltd, Mumbai, India. All other reagents were of analytical grade.

Preparation of STV microspheres

Batches of microspheres were prepared by ionotropic gelation procedure that concerned response between SA and polyca tionic particles like metallic component to supply a gel system of metallic component alginate.¹¹ SA and furthermore the m ucoadhesive substance compound were conveyed in sublimated water (10ml) to make a homogenized concoction compou nd blend. The API, STV (100mg) were intercalary to the substance compound in mixture and blended absolutely withast irrertomake agooeyscattering. The following scattering was then intercalary through a 22G needle into salt (4% w/v). The expansion was done persistent blending at 200rpm. The intercalary beads were kept up inside the salt response for half- hour to complete the hardening response and to supply unbending round microspheres.¹²

Table I. Formulation table of STV microspheres

Formulation code	Drug : Polymer ratio	Polymer blend ratio
MF1	1:2.5	SA : GG (1.5:0.5)
MF2	1:3	SA : GG (2:1)
MF3	1:3.5	SA : GG (2.5:1)
MF4	1:4	SA : GG (3:1)
MF5	1:2.5	SA : LG (1.5:0.5)
MF6	1:3	SA : LG (2:1)
MF7	1:3.5	SA : LG (2.5:1)
MF8	1:4	SA : LG (3:1)
MF9	1:2.5	SA : XG (1.5:0.5)
MF10	1:3	SA : XG (2:1)
MF11	1:3.5	SA : XG (2.5:1)
MF12	1:4	SA : XG (3:1)

Characterization of prepared microspheres

Fourier transform-infrared spectroscopy (FT-IR)

Fourier transform infrared (FT-IR) Spectroscopy was used to study the physical and chemical interaction between the drug and excipients used. FT-IR spectra of pure drug and optimized draft forming floating matrix tablet were recorded using KBr mixing method on FT-IR Spectrophotometer (FT-IR-1700, Shimadzu, Tokyo, Japan)¹³.

Differential scanning calorimetry (DSC)

DSC was used to study physical and chemical interaction between the drug and excipients used. DSC spectra of pure drug and drug composite mixture were recorded on differential scanning calorimeter (DSC-60, Shimadzu, Tokyo, Japan)¹⁴.

Particle size distribution

The particle size distribution of microspheres was determined using sieving method as described by Yüce and Canefe.¹⁵ Weighed microspheres of each formulation were put in a set of sieves fixed on the universal drive unit (Erweka, AR402, Germany).

Percentage yield of microspheres

The production yield of microspheres of each batch was calculated by using the weight of the final product after drying with respect to the initial total weight of the drug and polymer used for preparation of microspheres, and the percentage product on yield was calculated using.¹⁶

$$\text{Yield\%} = \frac{\text{Total weight of microspheres}}{\text{Total weight of drug and polymer}} \times 100$$

Drug entrapment efficiency

Drug entrapment efficiency (DE) was determined using the method described by Garg and Gupta. Accordingly, accurately weighed microspheres equivalent to 40 mg of drug was stirred with 100 mL of pH 1.2 (0.1 N HCl) for 2 h.¹⁷ The solution was filtered and after suitable dilution drug content was determined spectrophotometrically. The amount of drug entrapped in the microspheres was calculated by the following formula:

$$\text{DE\%} = \frac{\text{Mass of drug in microspheres}}{\text{Mass of drug in formulation}} \times 100$$

Scanning electron microscope (SEM) study

Scanning electron microscopy (JEOL 5400, Tokyo, Japan) was utilized to decide shape, surface geography and composition and look at the morphology of broken or separated surface. Samples spread on the little square plate and covered with a gold particle for 5-6 min.¹⁸ The ready test was kept inside the chamber and pictures caught with various amplifications.

In vitro buoyancy

Microspheres containing 300 mg of STV were spread over the surface of USP Type II (paddle) dissolution apparatus (Disso 2000, India) filled with 900 ml of 0.1 N HCl for 2 h followed by pH 6.8 phosphate buffer at $37 \pm 0.5^\circ\text{C}$ and agitated with a paddle rotating at 100 rpm. At the end of this period, the layer of buoyant particles on the surface of the medium was collected and the sinking particulates were separated by filtration.¹⁹ Both particle types were dried overnight in an oven drier (Riddi Pharma, India) at 40°C . Dried weights were measured, and buoyancy was determined by the weight ratio of the floating particles to the sum of floating and sinking particles.

$$\% \text{ Buoyancy} = \frac{\text{Dry weight of floated microspheres}}{\text{Total weight of floated and settled microspheres}} \times 100$$

In vitro drug release study

A USP Type II (paddle) dissolution apparatus (Disso 2000, India) was used to study *in vitro* drug release from microspheres as described elsewhere. Accordingly, an amount of the microsphere equivalent to 40 mg of STV filled in a hard gelatin capsule (size 0) was placed in the dissolution medium containing 900 ml of 0.1 N HCl for 2 h followed by pH 6.8 phosphate buffer at $37 \pm 0.5^\circ\text{C}$ with paddle rotating at 100 rpm.²⁰ Samples of 5 ml were withdrawn at 0.5, 1, 2, 4, 6, 8, 10 and 12 h and filtered. An equal volume of the dissolution medium was replaced every time in the vessel after each withdrawal to maintain sink condition. Each of the sample solutions were analyzed spectrophotometrically for the drug content at 274 nm.²¹ From this, the percentage of drug release was calculated and plotted as a function of time to study the pattern of drug release.

Kinetic modelling of drug release

To analyze the mechanism of drug release from the microspheres the *in vitro* dissolution data was fitted to zero order, first order, Higuchi's release model and Korsmeyer-Peppas model.^{22,23}

Stability study

To determine the stability study of the gastro-retentive microspheres of STV were packed in 40 ml HDPE bottle and stored at $40 \pm 2^\circ\text{C}$ and $75\% \pm 5\%$ RH for a period of six months as per the ICH guidelines.²⁴ The microspheres were withdrawn at a period of 1, 3 and 6 months and evaluated for content uniformity and dissolution study.

In vivo studies

A prior approval of institutional animal ethical committee was obtained from Sanzyme Pvt Ltd., Hyderabad, India (Regn. No. 1688/PO/E/2013/CPCSEA, Dated 16-Aug-2018) for carrying out this experiment. Male New Zealand white rabbits weighing between 1.5 to 2.0 kg were used for *in vivo* studies of orally administered STV. Animals were housed at $25 \pm 1^\circ\text{C}$ in air conditioned room at a relative humidity of $60 \pm 5\%$ and were provided with water and

ndard rabbit feed obtained from Sanzyme Pvt Ltd., Hyderabad. Animals were fasted for 24 h prior to the administration of the pellet formulation, but had free access to water. The microsphere formulation MF4 loaded capsules were administered orally with a soft plastic tube to different groups of rabbits separately. Drug solution was also administered orally with a soft plastic tube (STV 5 mg/rabbit) to another group of animals. About 2 ml of blood was collected at 0, 0.5, 1.0, 2.0, 3.0, 4.0 and 6.0 h after oral solution administration and 0, 1, 2, 4, 6, 8 h after tablet formulation administration from marginal ear vein into heparinized tubes and plasma was separated immediately by centrifugation and frozen at -20°C . The plasma samples were analyzed for STV by HPLC method.²⁵

RESULTS AND DISCUSSION

Drug-excipient compatibility study

Drug-polymer compatibility studies were carried out using Fourier Transform infrared spectroscopy to establish any possible interaction of Drug with the polymers used in the formulation. The interaction study between the drug and polymer was evaluated using FT-IR spectrophotometer. The FT-IR spectra of the formulations were compared with the FT-IR spectra of the pure drug (Figure 1). There was no significant difference in the IR spectra of pure and drug loaded microcapsules. Four bands characteristic of O-H stretching, N-H stretching of secondary amine, C-H stretching and C=O stretching in the pure drug was unchanged in the prepared formulations.

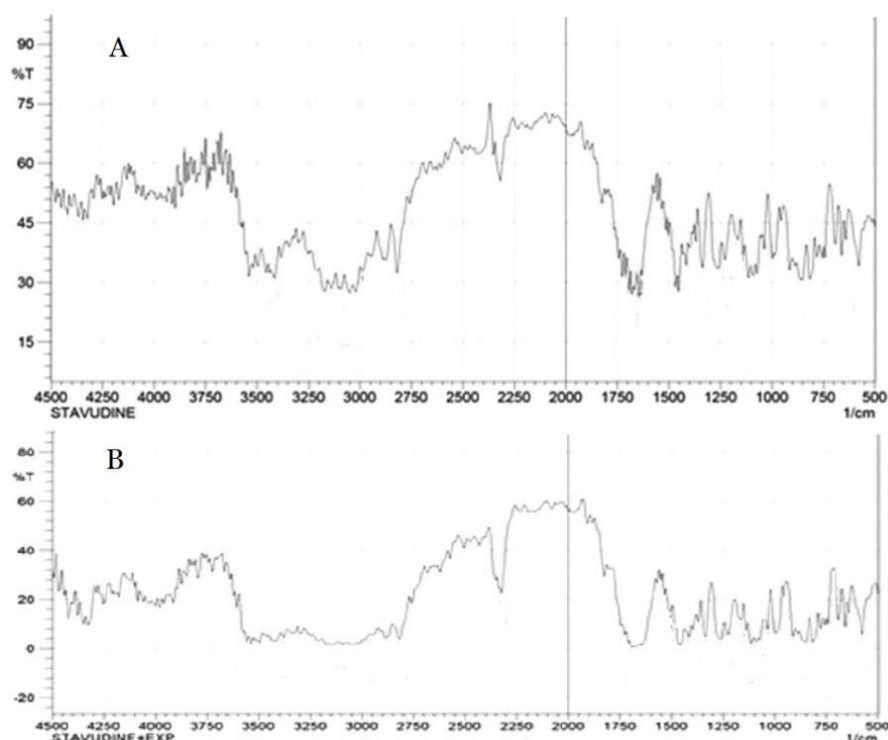


Figure 1: FT-IR spectra of (A) pure drug and (B) optimized formulation MF4

The DSC thermogram of STV showed sharp endothermic peak at 172.1°C , while that of formulation MF4 showed broad endothermic peak at 169.36°C (Figure 2). The DSC thermograms of a mixture of STV with SA-GG in formulations showed sharp endothermic peaks for STV at the temperature similar to that of the peak of STV alone. This indicated that there was no drug-excipient interaction in the formulations.

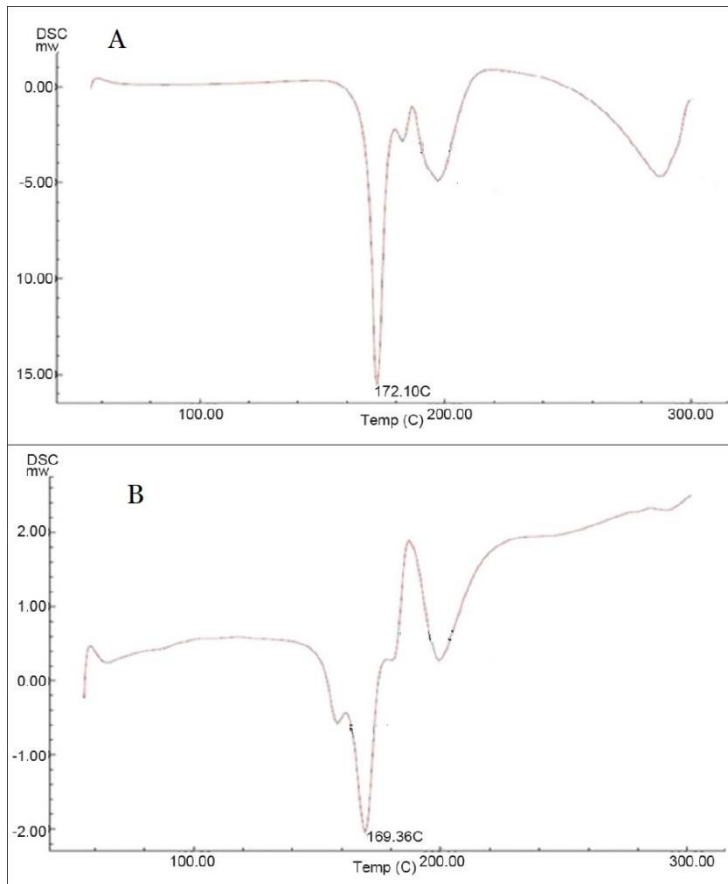


Figure 2: DSC thermograms of (A) STV and (B) optimized formulation (MF4)

Characterization of prepared microspheres

Scanning electron microscopy and particle size analysis showed that microspheres containing SA and GG as polymer blend had a size range of 512 to 826 μ , microspheres containing SA and LG as polymer blend exhibited a size range between 517 to 834 μ and microspheres containing SA and XG as polymer blend had a size range of 664 to 903 μ (Figure 3). The particle size as well as drug entrapment efficiency of the microspheres increased with increase in the polymer concentration.

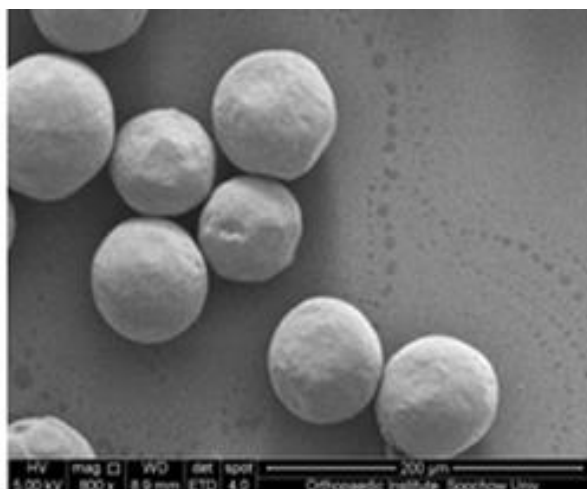


Figure 3: Surface morphology of optimised formulation MF4

The percent yield, drug content, drug entrapment, buoyancy, lag time and floating time of the prepared microspheres are summarized in Table 2. It was observed that as the polymer ratio in the formulation increased, the product yield and drug entrapment efficiency were also increased. The buoyancy was increased as the polymer concentrations were increased. The lag time and floating times were decreased as the polymer concentrations were increased.

Table II: The yield, drug content, drug entrapment, buoyancy, lag time and floating time of the prepared microspheres

S.N o.	Formulation code	Yield (%)	Drug content (mg)	Drug entrapment efficiency (%)	Buoyancy (%)	Lag time (min)	Floating time (h)
1	MF1	80±0.02	12.4±0.08	82.66±0.06	63±0.05	12	8
2	MF2	83.33±1.05	12.66±1.32	84.4±0.04	67±1.30	10	8
3	MF3	85±2.07	12.70±0.24	84.66±0.01	75±2.06	7	8
4	MF4	88±0.01	13.29±0.06	88.66±0.03	79±0.07	6	10
5	MF5	62.22±0.4	8.07±2.07	53.2±0.09	89±1.02	3	>12
6	MF6	80±1.04	8.25±0.38	55±0.04	85±0.82	3	>12
7	MF7	80±1.50	10.33±1.02	68.86±0.07	70±1.03	7	10
8	MF8	87±0.02	11.5±0.05	76.66±0.02	76±0.05	8	10
9	MF9	80±0.05	10.01±1.06	66.73±2.03	84±0.92	4	>12
10	MF10	86±0.04	10.5±0.18	70±1.04	78±1.09	7	11
11	MF11	86.66±0.07	11.25±1.26	75±0.45	74±2.06	8	11
12	MF12	87.5±0.03	11.88±0.09	79.2±0.02	81±0.05	6	>12

It was observed that with an increase in polymer concentration, the percentage of swelling also increased. Thus we can say that amount of polymer directly affects the swelling ratio. As the polymer to drug ratio increased, the percentage of swelling increased from 27 to 87% for microspheres containing SA and GG polymer blend, 23 to 67% for microspheres containing SA and LG polymer blend and 28 to 79% for microspheres containing SA and XG polymer blend.

In vitro drug release studies and kinetics

Formulation MF4 showed 94.66% release of the drug for a period of 12h with relatively rapid release when compared to other formulations prepared by using other polymers (Figures 4-6). The drug release profiles are fitted to release kinetics the slope for Peppas was in the range of 0.39 to 0.987 indicating both diffusion and erosion of both polymers.

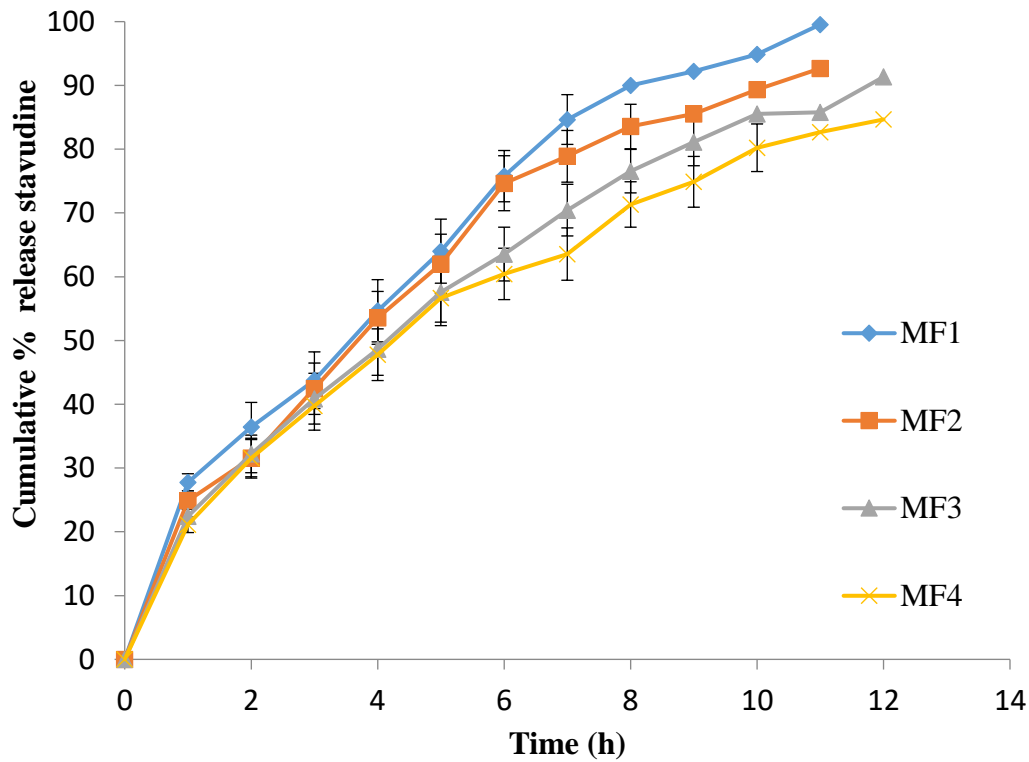


Figure 4: Comparison of *in vitro* drug release profile of STV microspheres containing SA and GG polymer blend

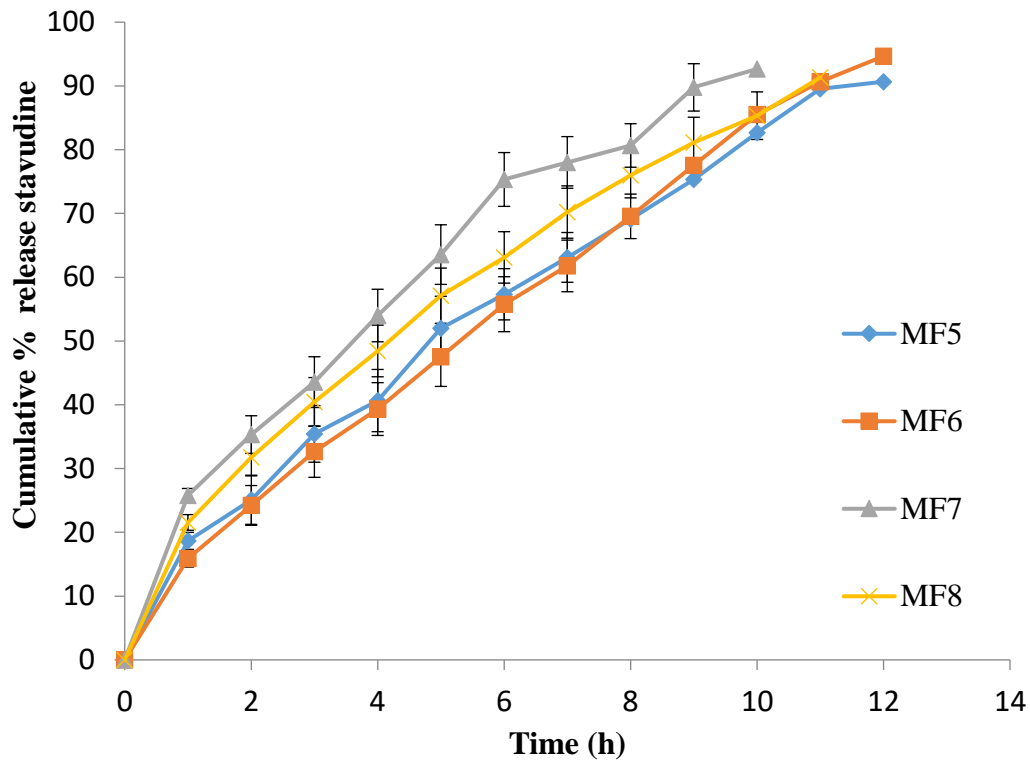


Figure5: Comparison of *in vitro* drug release profile of STV microspheres containing SA and LG polymer blend

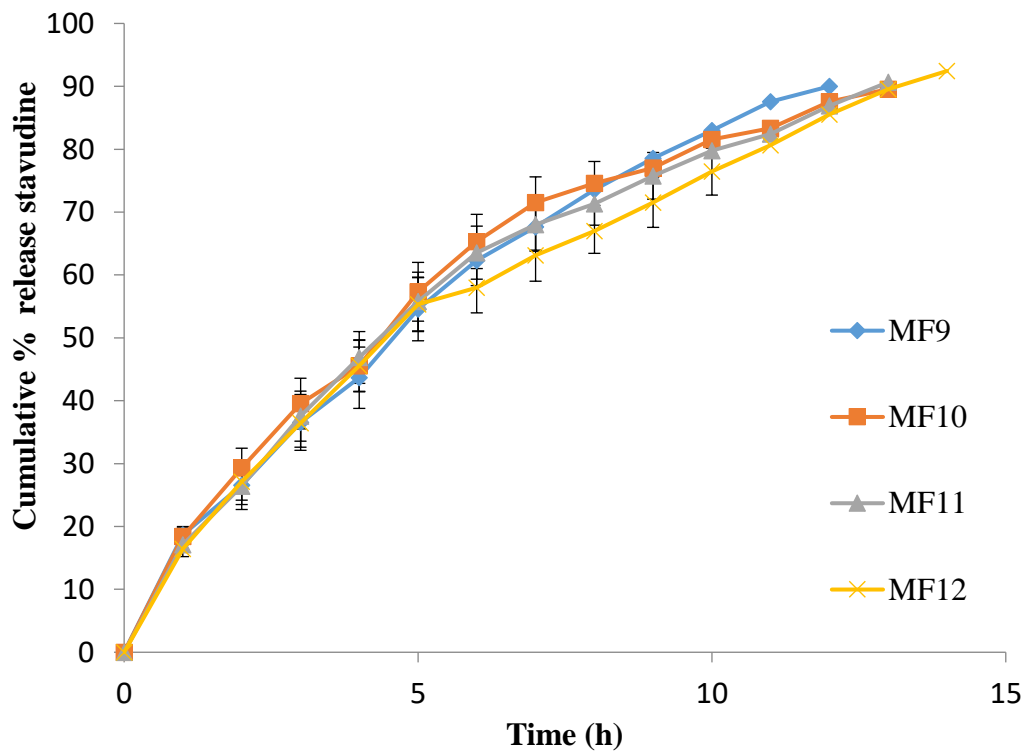


Figure6: Comparison of *in vitro* drug release profile of STV microspheres containing SA and XG

Table III: Release kinetics studies of the prepared formulations

Formulation	Release model								
	Zero order		First order		Higuchimatrix		Koresmeyer-Peppas		
	K	R ²	K	R ²	K	R ²	n	K	R ²
MF1	19.286	0.769	1.704	0.689	-0.56	0.891	0.534	1.251	0.883
MF2	14.076	0.88	1.772	0.859	-4.192	0.949	0.573	1.189	0.941
MF3	8.136	0.948	1.843	0.914	-9.213	0.954	0.651	1.096	0.949
MF4	5.12	0.962	1.899	0.883	-12.497	0.941	0.721	1.034	0.954
MF5	22.026	0.74	1.678	0.658	2.377	0.882	0.476	1.305	0.872
MF6	14.876	0.876	1.771	0.854	-3.58	0.95	0.557	1.209	0.939
MF7	12.216	0.908	1.799	0.954	-6.486	0.962	0.633	1.141	0.948
MF8	10.746	0.92	1.813	0.96	-7.834	0.963	0.668	1.104	0.949
MF9	20.886	0.755	1.69	0.673	1.089	0.888	0.504	1.281	0.883
MF10	14.416	0.878	1.773	0.854	-4.018	0.949	0.569	1.197	0.94
MF11	10.186	0.929	1.817	0.943	-7.887	0.961	0.645	1.116	0.951
MF12	9.626	0.931	1.842	0.951	-9.233	0.96	0.69	1.089	0.953

Stability studies for the optimised formulation

Stability studies showed that there was no significant reduction in the drug content (<10%0.7%) and hence the optimized formulation was found to be stable. By observing the stability studies it is concluded that the optimized formulation is stable throughout the entire period of 3 months and the drug release profile is also intact throughout the time being.

In vivo studies on optimized formulation

The results of *in vivo* studies are shown in Figure 8. The *in vivo* studies showed that the optimized microspheres of STV showed a prolonged drug release pattern as compared to oral solution of the STV. The formulation showed C_{max} of 89 ng/mL, T_{max} of 8 h, $t_{1/2}$ of 3.68 h, K_{el} of 0.89/h, $AUC(0-t)$ of 2127 ng-h/mL, MRT of 10.57 h and $AUMC(0-t)$ 23765 ng-h²/mL. The kinetic pattern was shown in Figure 7.

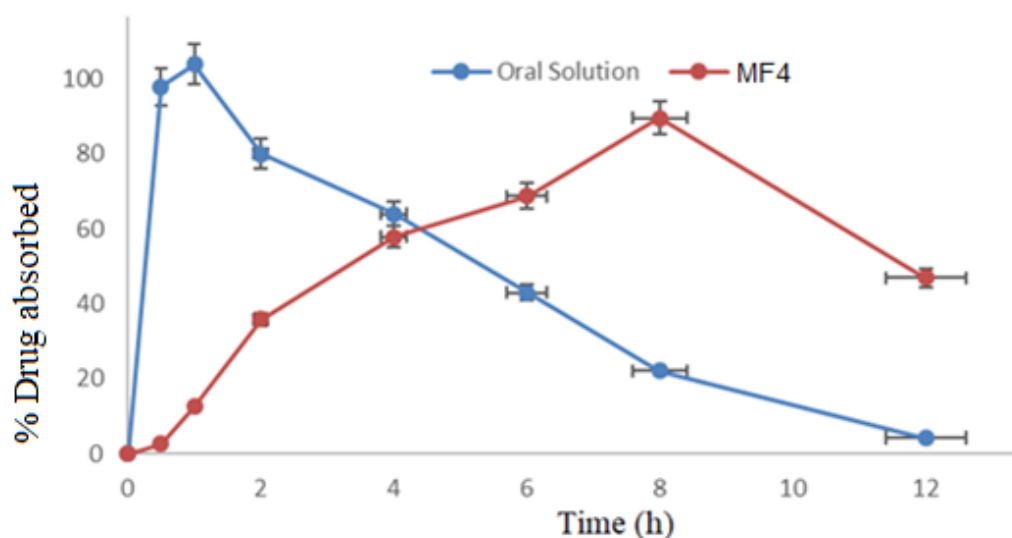


Figure 7: Plasma concentration (ng/ml) of STV oral drug solution and gastro-retentive microspheres formulation MF4

The objective of the study is to formulate and evaluate STV floating microspheres by ionotropic gelation method. The preparation contained 12 formulations using different polymers, i.e. XG, GG and LG in different ratios. The prepared batches of STV floating microspheres were evaluated like *in vitro* buoyancy, swelling index, drug entrapment efficiency and *in vitro* release studies and *in vivo* studies. *In vitro* release studies were done in order to investigate the release of drug from floating alginate microspheres. STV was released rapidly from alginate-cellulose microspheres in the gastric fluid. In the formulations with LG and XG as copolymers maximum drug release occurred 92 and 90% respectively. In the absence of gas forming the release rate was slow. This can be explained as the highly dense internal structure of the alginate microspheres. *In vivo* results of STV microspheres showed the T_{max} of 8h, C_{max} of 89 ng/mL, and $AUC_{(0-t)}$ of 2127 ng-h/mL.

CONCLUSION

Gastro-

retentive floating microspheres of STV using SA and GG, LG, XG as polymer blend were formulated to deliver stavudine via oral route. The prepared formulations were subjected to various evaluation parameters and the optimized formulation selected based on the drug release and MF4 is selected as the best formulation for consistent drug release in the given time. Microspheres resulted in significant improvement in physical properties of the drugs especially with respect to flow properties. *In vivo* results showed increased bioavailability of the optimized formulation which will be reducing the dose and side effect of stavudine. Therefore the Formulation MF4 could be successfully used as floating drug delivery system for stavudine.

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REFERENCES

1. Mahale MM, Saudagar RB. Microsphere: A Review. JDDT. 2019;9(3-s):854–856. doi: 10.22270/jddt.v9i3-s.2826.
2. Deveswaran R, Manavalan R, Madhavan V, Bharath S. Formulation and evaluation of albumin microspheres containing aceclofenac. Int J Pharm Sci Rev Res. 2010; 4 (1): 112–117.
3. Vinukonda A, Kunderu R, Gunnam S. Review on mucoadhesive microspheres. Int J ChemTech Res. 2018; 11: 277–289.
4. Thummar AV, Kyada CR, Kalyanvat R, Shreevastva BA. Review on mucoadhesive microspheres as a novel drug delivery system. Int J Pharm Scholar 2013; 2: 188–200.
5. Patel NR, Patel DA, Bharadia PD, Pandya V, Modi D. Microsphere as a novel drug delivery. Int J Pharm Life Sci. 2011; 2(8): 992–997.
6. Padala NR, Prakash K, Bonepally CSR, Krishnaveni B, Shanta KK, Lakshmi NM. Stavudine loaded microcapsules using various cellulose polymers: preparation and in-vitro evaluation. Int J Pharm Sci Nanotech. 2009; 2(2): 551–556.
7. Mahendar R, Kranthi Kumar G, Chinna RP, Rao Patnaik KS. Formulation, Optimization and in vitro characterization of stavudine gastro retentive floating matrix tablets. Int J Pharm Sci Drug Res. 2016; 8 (3): 128–135.
8. Vohra SY, Patil CC. Development and characterization of stavudine microspheres prepared using different polymers. J Pharm Res. 2009; 2 (3): 953–957.
9. Srikrishna T, Gobinath M, Venkata AP, Sai GM, Sudheer S. Study of stavudine multiparticulate floating drug delivery system prepared by emulsion gelation technique. Saudi J Med Pharm Sci. 2017; 3: 714–727.
10. Sundaramoorthy R, Vuyyuru M, Dhanaraju MD. Formulation and evaluation of stavudine loaded sodium alginate beads by ionotropic gelation method. Int Res J Pharm 2014; 5(9): 706–712.
11. Choudhury PK, Kar M. Preparation of alginate gel beads containing metformin hydrochloride using emulsion - gelation method. Trop J Pharm Res 2005; 4 (2): 489–493.
12. Parul T. Preparation and characterization of aceclofenac microspheres. Asian J Pharm. 2008; 1: 110–115.

13. S.Tiwari and B Mishra. Multilayered membrane-controlled microcapsules for controlled delivery of isoniazid. *DARU J Pharm Sci.* 2011; 19 (1): 41–46.
14. Rahulnair, Arun KKS, Vishnupriya KV. preparation and characterization of Rizatriptan loaded solid lipid nanoparticles. *J Biomed Sci Res.* 2011; 3: 392–396.
15. Yüce M, Canefe K. Indomethacin-loaded microspheres: preparation, characterization and in-vitro evaluation regarding ethylcellulose matrix material. *Turk J Pharm Sci.* 2008; 5 (3): 129–142.
16. Nochosa A, Douroumis D, Bouropoulos N. In vitro release of bovine serum albumin from alginate/HPMC hydrogel beads. 2008; 74 (13): 451–457.
17. Prabu LS, Shirwaikar AA. Formulation and evaluation of sustained release microspheres of rosin containing aceclofenac. *Ars Pharm.* 2009; 50 (2): 51–62.
18. Young R, Chulsoo S. Preparation and characterization of alginate–carrageenan complex films. *J ApplPolym Sci.* 2006; 99 (6): 3483–3490.
19. Arrora S, Ali J, Khar RK, Baboota S. Floating drug delivery systems: A review. *AAPS Pharm Sci Tech.* 2005; 6: 372–390.
20. Pradeesh T, Sunny M, Varma H, Ramesh P. Preparation of microstructured hydroxyapatite microspheres using oil in water emulsions. *Indian Acad Sci.* 2005; 28 (5): 383–390.
21. Khadeer ZS, Srinivasulu Y, Pushpa LE. Development and validation of new spectrophotometric method for the estimation of stavudine in bulk and pharmaceutical dosage form. *Sch. Acad. J. Pharm.,* 2013; 2(4):319-322.
22. Higuchi T. Mechanism of sustained-action medication. Theoretical analysis of rate of release of solid drugs dispersed in solid matrices. *J Pharm Sci.* 1963; 52: 1145–1149.
23. Fundeanu G, Esposito E, Mihai D. Preparation and characterization of alginate Microspheres by a new emulsification method. *Int J Pharm.* 1998; 170: 11–21.
24. Kharia AA, Hiremath SN, Singhai AK and Jain SK: Design and optimization of floating drug delivery system of acyclovir. *Indian J Pharm Sci.* 2010; 72 (5): 599–606.
25. Surajpalverma, bioanalytical method development and validation for the simultaneous estimation of lamivudine and stavudine in human plasma by HPLC, *ActaPoloniaePharmaceutica Drug Res.* 2010; 67: 429–437.